

Amendments to the Specification:

Please amend the specification as indicated below.

Please replace paragraph [0034] of the application (paragraph [0032] of pre-grant publication 20060106100) with the following amended paragraph:

[0034] In a specific embodiment of the invention, the obtained cyclopalladated compounds are represented in the Schemes 3 and 4 below, respectively derived from N,N-dimethyl-benzylamine (scheme 3) and from ~~alkines~~ alkynes pyridinyl-phenyl-~~ethine~~ ethyne (scheme 4A) and 1-phenyl-3-N,N-dimethylamine-~~propine~~ propyne (scheme 4B).

Please replace paragraph [0071] of the application (paragraph [0064] of pre-grant publication 20060106100) with the following amended paragraph:

[0071] FIG. 16 presents the in vivo antitumor activity of compounds analogue to drug 1, now containing functionalized ~~alkines~~ alkenes derived from functionalized alkynes as cyclometalling agent.

Please replace paragraph [00130] of the application (paragraph [0115] of pre-grant publication 20060106100) with the following amended paragraph:

[00130] The intensities of fluorescence signals were monitored by a thermostatic spectrofluorimetry Hitachi F-2000. Wavelength was calibrated at 380 nm for

excitation and 440 nm for emission. The activation of the enzyme was made by incubation for five minutes at 37°C in a buffer solution of 50 mM sodium phosphate (pH 6.4) containing NaCl 200 mM, EDTA 1 mM and DTT 2 mM. Measurements were taken in the same activation buffer of cathepsin B and kinetic standards were set up by taking the initial ~~hydrolysis~~ hydrolysis rate under various concentrations of the substrate in the presence or absence of different concentrations of organometal compounds. The results were analysed by non-linear regression by using the software GraFit 3.01 (Erithacus Software Ltd.).

Please replace paragraph [00131] of the application (paragraph [0116] of pre-grant publication 20060106100) with the following amended paragraph:

[00131] The kinetic model of Scheme 1 describes the effect of heparin over the ~~hydrolysis~~ hydrolysis of Z-Phe-Arg-MCA by cathepsin B:

Please replace paragraph [00134] of the application (paragraph [0118] of pre-grant publication 20060106100) with the following amended paragraph:

[0134] Kinetic studies were made for the cyclopalladated complex [Pd (C²,N-(R+dmpa)(dppf)N₃]. As shown by the FIG. 1A, the presence of organometal in the kinetic test of Cathepsin B results in a reduction of the values k_{cat} for the ~~hydrolysis~~ hydrolysis of Z-Phe-Arg-MCA. On the other hand, Figure 1B shows that the compound also notably increases the affinity of Cathepsin B by the substrate Z-Phe-Arg-MCA. The effect of organometal over the activity of endopeptidase Cathepsin B can be described by means of a mixed hyperbolic curve with inhibition standard as

shown by Scheme 1. The efficiency of the substrate ~~hydrolysis~~ hydrolysis system can be changed by modifications in K_s (parameter α) or V_{max} (parameter β). Data were treated according to Equation 1, by using non-linear regression and the values for the constants were calculated. Results show that $[Pd(C^2,N-(R+dmpa)(dppf)]N_3$ is linked to free Cathepsin B (E) with dissociation constant $K_H = 12 \pm 1 \mu M$ and the compound is linked to the enzyme-substrate (ES) complex with dissociation constant $\alpha K_H = 2.4 \pm 0.3 \mu M$. The complex also induced a 5.3-time increase of the affinity between Cathepsin B and the substrate Z-Phe-Arg-MCA; K_s value was reduced from 110 ± 15 to $21 \pm 2 \mu M$ in the presence of the organometal compound, $\alpha = 0.19 \pm 0.02$ (Fig. 1B), while the value of k_{cat} in the presence of $[Pd(C^2,N-(R+dmpa)(dppf)]N_3$ was also reduced 5.6 times, $\beta = 0.18 \pm 0.02$. The cyclopalladated compound reduced the product formation constant to 36 ($\beta = 0.18 \pm 0.02$) in the same proportion increasing the affinity of Cathepsin B by the substrate Z-Phe-Arg-MCA ($\alpha = 0.19 \pm 0.02$), i.e., $\alpha = \beta$. Despite Cathepsin B having been strongly inhibited by the organometal complex (81% inhibition), its efficiency for that substrate in the presence of the organometal complex was not changed, $\beta/\alpha = 1.1 \pm 0.1$. The ~~hydrolysis~~ hydrolysis rate of the second order substrates was the same, $k_{cat}/K_s = 4.5 \cdot 10^5 M^{-1}s^{-1}$, in the presence or absence of $[Pd(C^2,N-(R+dmpa)(dppf)]N_3$.

Please replace paragraph [00135] of the application (paragraph [0119] of pre-grant publication 20060106100) with the following amended paragraph:

[00135] Cathepsin B and other cysteine-proteases pertaining to the papain superfamily contain highly conserved folding structure (Turk, V.; Bode, W., *Lysosomal cysteine proteinases and their inhibitors cystatins. Innovations in Proteases and their Inhibitors*. Aviles, F. X. (Ed.), Walter de Gruyter & Co., Berlin,

Germany, 1993). These enzymes are linked to peptide substrates and use the pair of ions thiolate-imidazolium to act in its proteolytic activity. This association between the cysteine residue and histidine grants high nucleoficity to the active site of the cysteine (Michaud, S.; Gour, B. J., *Cathepsin B inhibitors as potential anti-metastatic agents*. Exp. Opin. Patents., 1998, 8, 645-672). The rupture of amide linkages of the substrate involves the formation of an intermediate acyl enzyme. After the formation of the non-covalent complex of Michaelis, the active site thiolate attacks the peptide linkage to form an oxanion which is stabilized in the so-called "oxanion channel" by a glutamine residue. The collapse of the tetrahedric intermediate results in the acyl enzyme and releases product. Subsequently, the ~~hydrolysis~~ hydrolysis of the acyl enzyme regenerates the catalytic ionic pair and releases the new product, i.e. carboxylic acid. Example 5 shows enzymatic inhibition assays by means of the compounds of the invention.

Please replace paragraph [00187] of the application (paragraph [0171] of pre-grant publication 20060106100) with the following amended paragraph:

[00187] The compounds of the invention can also be coupled to soluble polymers as possible drug carriers. These polymers can include ~~polyvinylpyrrolidone~~ polyvinylpyrrolidone, ~~piran~~ pyran copolymer, ~~poly-hydroxypropylmethacrilamidephenol~~ poly-hydroxypropylmethacrylamidephenol, poly-hydroxyethylaspartamidephenol or polyethylene oxide-polylysine ~~oxide-~~ polylysine substituted with palmitoyl residues. Furthermore, the compounds of the invention can be coupled to a class of useful biodegradable polymers to reach the controlled release of a drug, such as polylactic acid, polyglucolic acid, copolymers of polylactic and polyglucolic acid, polyepsilon caprolactone, poly-hydroxy butyric acid,

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polyorthoesters, polyacetals, ~~polydi-hydropirans~~ polydi-hydropyrans,
~~polycyanoacilates~~ polycyanoacylates and crosslinked or amphypathic block
copolymers of hydrogels.

Please replace paragraph [00319] of the application (paragraph [0289] of pre-grant
publication 20060106100) with the following amended paragraph:

[00319] *In vitro* assays were made to verify the cytotoxic effect of cyclopalladated
compounds of murine melanoma B16F10-Nex2 cells. The cells were incubated for 24
hours under different concentrations. Cell viability was determined by an assay
with MTT. As shown in FIG. 13, 100% efficiency is observed under concentration
about 1.25 μ M of the drugs containing ethane biphosphine (drugs 1, 2 and 3). The
same efficiency is noted in the drug containing bis-diphenylphosphine-ferrocene, but
under higher concentration, 10 μ M (drug 6) and for cyclopalladated compounds
containing functionalized ~~alkines~~ alkenes derived from functionalized alkynes (drug
10), about 20 μ M. Figure 13 shows the described cytotoxic effect, in which the
reduction percentuals of cell viability, in comparison to a control with cells with no
drugs, are represented over the bars.

Please replace paragraph [00324] of the application (paragraph [0294] of pre-grant
publication 20060106100) with the following amended paragraph:

[00324] Drugs with functionalized ~~alkines~~ alkenes derived from functionalized alkynes were also tested *in vitro* (Figure 16) following the same protocol and presented cytotoxic effect of 100%, about 100 μ M. This effect is represented in the following figure, also showing that, among them, drugs 11 and 14 presented significant inhibition from 40 to 50%, about 1 μ M.

Please replace paragraph [00325] of the application (paragraph [0295] of pre-grant publication 20060106100) with the following amended paragraph:

[0295] Result analysis offers some important co-relations between chemical structure and drug activity. 100% efficiency is observed under a concentration of about 1.25 μ M of the drugs containing ethane biphosphine (drugs 1, 2 and 3). The same does not occur for cyclopalladated compounds containing functionalized ~~alkines~~ alkenes derived from functionalized alkynes (drug 10), in which a concentration about 20 μ M is required. From all tested drugs, we can notice that drugs 1 and 6 were those presenting an inhibitory effect over tumor development of about 60% over other drugs. We can see the drug 1 as being the most efficient one to inhibit tumor development, but test animals treated with drug 6 present a lower number of deaths. From the complexes containing functionalized ~~alkines~~ alkenes, drugs 11 and 14 are those presenting significant inhibition, from 40 to 50%, of about 1 μ M.